



































Time Average definition The system of the cycle of oscillation to compute an energy loss rate of the oscillating atoms:

$$\langle U \rangle = \langle P.E. \rangle_{\text{avg.}} + \langle K.E. \rangle_{\text{avg.}} \\ = \frac{1}{2} k \langle x^2 \rangle_{\text{avg.}} + \frac{1}{2} m \langle \frac{dx}{dt} \rangle_{\text{avg.}} \\ = \langle U_0 \rangle e^{-\gamma t}$$

19

Energy Loss

$$\gamma = \left| \frac{1}{\langle U \rangle} \frac{d \langle U \rangle}{dt} \right| = \gamma_{\rm rad} + \gamma_{\rm m}$$

- Here γ is the energy loss rate which we have explicitly broken out into radiative and non-radiative terms.
- The radiative part is due to spontaneous emission.
- The non-radiative energy loss term is due to inelastic collisions with other atoms, walls, etc. In solids this loss is due to a coupling of the energy into the lattice.

20











Dephasing Time	
• Initially, all the dipoles are oscillating in phase	
${p}_{x_0}=N_0\mu_{x_0}$	
• At <i>t</i> > 0, we have a decreasing number of dipoles that have suffered collisions.	not
$p_x(t) = N(t)\mu_x(t)$	
• If collisions occur at a random rate of $1/T_2$ collisions per atom per second. Then the decay of uncollided atoms $N(t)$ is given by	
$\frac{dN(t)}{dt} = -\frac{N(t)}{T_2} \Longrightarrow N(t) = N_0 e^{-\frac{t}{T_2}}$	
	26

Polarization Now assume we can align all of the dipoles in our unit volume at t = 0 and let them oscillate (at t = 0, let the external field go to zero). We find that p as a function of time

$$p(t) = p_0 e^{-\frac{\gamma}{2}t} e^{j\sqrt{\omega_0^2 - \left(\frac{\gamma}{2}\right)^2}t} e^{-\frac{t}{T_2}}$$
$$= p_0 e^{-\left(\frac{\gamma}{2} + \frac{1}{T_2}\right)t + j\sqrt{\omega_0^2 - \left(\frac{\gamma}{2}\right)^2}t}$$

 γ : Energy loss rate

 $1/T_2$: Dephasing rate

27